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The Crystal Structure of TiFeSi and Related Compounds

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TiFeSi crystallizes with an orthorhombic unit cell of space group Ima2 (C_{2v}^{2r}) with lattice constants $a=6.997\pm0.002$, $b=10.830\pm0.005$, $c=6.287\pm0.002$ Å. The structure is of a new type, which is a superstructure of the ordered Fe₂P-type, with 12 formula units in the orthorhombic unit-cell. TiFeSi has a strong tendency to form pseudohexagonal multiple twins with the common *a* axis as twin axis. The structure has been refined by the least-squares method from single-crystal X-ray data, giving a final *R* value of 0.094 for the 344 observed reflections. TiFeGe crystallizes with the same structure: $a=7.155\pm0.002$, $b=11.025\pm0.007$, $c=6.405\pm0.003$ Å, while TiCoGe is of the ordered Fe₂P-type: $a=6.222\pm0.002$, $c=3.7267\pm0.0010$ Å. The coordination polyhedra in the structures TiFeSi and TiCoGe are very similar to those occurring in TiNiSi (*E* phase) and Nb₅Cu₄Si₄. The interatomic distances in these structures are discussed.

Introduction

Recent investigations on ternary systems with two transition metals and silicon or germanium showed the existence of a large number of isotypic phases at the composition 1:1:1 which are called *E* phases (for a summary of references see Jeitschko, 1968; Jeitschko, Jordan & Beck, 1969), the structure of which was determined for the prototype TiNiSi to be of the ordered PbCl₂-type (Shoemaker & Shoemaker, 1965). In one of these studies (Spiegel, Bardos & Beck, 1963) a

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phase of composition TiFeSi was observed (Spiegel, 1963) which was not isotypic with TiNiSi. This phase was also found independently by Freundlich & Mochai (1966) and Markiv, Gladyševeskij, Skolozdra & Kripjakevič (1967). The latter authors also gave unit-cell data (6/mmm, a=6.24, c=6.96 Å) which are closely related to the unit cell found in the present investigation; they also reported an experimental density of 5.57 g.cm⁻³ for TiFeSi.

The crystal structure of TiFeSi seemed of interest in order to make a comparison with the E phase structure TiNiSi, especially since the ratios of the atomic sizes in TiFeSi are well within the limits found for the occurrence of E phases and therefore cannot be responsible for the change in structure.

Experimental

Samples of TiFeSi were prepared by arc melting the components (all of purity higher than 99.9%) under argon (99.996% Ar) atmosphere. One sample was annealed in an evacuated fused-silica tube at 1000°C for five days. Metallographic examination showed that the sample was about 99% single phase. Powder patterns taken with Cu Ka radiation in a Guinier-de Wolff camera (using one film as filter to absorb fluorescent radiation) gave the same pattern for both the annealed and the unannealed specimen. 'Single crystals' isolated from the crushed samples (annealed and unannealed) showed a strong tendency to twinning, which was not recognized immediately. Single-crystal photographs of such a specimen were taken with a Weissenberg (Cr $K\alpha$ and Mo $K\alpha$ radiation) and a Buerger precession camera (Mo $K\alpha$ radiation) and could be indexed with a hexagonal unit cell of a=6.29 and c = 3.50 Å.

A preliminary intensity calculation gave very good agreement for a structure proposal assuming an ordered Fe₂P-type (Rundqvist & Jellinek, 1959). However, the presence of a number of weak reflections required a doubling of the c axis in agreement with the unit cell found by Markiv *et al.* (1967).

The reciprocal lattice plane (hk1) of this larger hexagonal pseudocell, which contains only the weak superstructure reflections (Fig. 1) suggested additionally a doubling of the *a* axis. A reciprocal lattice based on this unit cell, however, contains a high number of regular extinctions which could not be accounted for by space group extinctions. Therefore, the possibility of twinning (Buerger, 1960) was investigated.

The metallographic examination of a polished and etched (3% HF, 3% HNO₃, 94% H₂O) specimen using polarized light clearly showed twin boundaries which were not visible under unpolarized light, in addition to normal grain boundaries. The twin boundaries were present both in the annealed and unannealed conditions. Furthermore, the inspection of the pseudohexagonal reciprocal lattice plane (hk1) mentioned above (Fig. 1), contained some higher angle reflections which were out of register by up to 0.5 mm. Thus, twinning was clearly established.

The metallographic examination with polarized light revealed that single crystals of dimensions as large as 0.3 mm in diameter were present in the specimens. However, a search for a non-twinned crystal in the crushed samples was not successful.* The closest approximation to a single crystal was a plate-shaped specimen with dimensions $0.20 \times 0.15 \times 0.04$ mm. It contained about 88% of one orientation, while the other two orientations of this multiple twin (trilling) comprised 5 and 7%. These percentages were estimated from the intensity-distribution in the pseudo-hexagonal reciprocal lattice plane (hk1) (Fig. 1). The Buerger precession photographs of this specimen showed orthorhombic symmetry for the superstructure reflections, whereas the reflections for the Fe₂P-like subcell still had the pseudohexagonal symmetry. With the lattice constants obtained from these photographs the Guinier-de Wolff powder pattern was indexed and a least-squares fit of the resolved lines (Table 1) gave the following lattice constants using silicon (99.98 Si, $a = 5.4311 \pm 0.0004$ with $\lambda Cu K\alpha_1 = 1.54051$ Å) as standard:

$$a = 6.997 \pm 0.002 \text{ Å} \qquad c = 6.287 \pm 0.002 \text{ Å} b = 10.830 + 0.005 \qquad V = 476.4 \pm 0.5 \text{ Å}^3.$$

In this setting the orthorhombic a axis corresponds to the twin axis which is also the pseudohexagonal axis of the Fe₂P-like subcell.

For the refinement of the structure intensities were estimated visually with the aid of an intensity strip, which was scaled logarithmically in steps of 10%. The reciprocal lattice planes (0kl) to (3kl) and (h, 2l, l)were recorded from the 88% single crystal mentioned above with a Buerger precession camera (Mo K α radiation). All symmetry related intensities were measured and averaged after division by the Lorentzpolarization factor. No absorption correction was made. The presence of about 12% of the other two orientations in the specimen did not affect the accuracy

*It is not known whether crushing the brittle samples produced additional twinning by deformation.



Fig.1 Three interpenetrating orthorhombic reciprocal lattice planes (1kl) of a trilling of TiFeSi, suggesting hexagonal symmetry.

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Table 1. Evaluation of Guinier-de Wolff powder patterns of TiFeSi and TiFeGe

Cu $K\alpha$ radiation

	TiFeSi			TiFeGe			
hkl* hkl	ā _c (Å)	I _c	I.	d _c (Å)	I _c	I _o	
110	5.8771	10	vvvw	6.0019	2	-	
$100 \begin{cases} 011 \\ 020 \end{cases}$	5.4372	48 21 J	w,d	5.5382	2/ 9}	vw	
121	3.5393	8	vvvw	3.6080	1	-	
130	3.4985	40 3	-	3.2690	1	- -	
110 ∫ 002	3.1435	20]	w.d	3.2025	,73}	m,d	
[21]	3.1306	37 J 45 1		3.18/6	54]	-	
101 {220	2.9386	26 }	w,d	3.0009	23	w	
112 1022	2.7719	1	-	2.8254	91	-	
200 { 040	2.7075	8)	vw,d	2.7562	3	vvw,a	
141 1202	2.3431	302	vvs.d	2.3868	180 }	vs,d	
111 231	2.3329	689		2.3799	456		
310 132	2.2801	25 20	vvw vvw	2.3311	24 25	vvw	
201 222	2.1467	614]	vvs.d	2.1898	704	vvs	
150	2.1412	263 J 28	vw	2.1834	296 J 37	vw	
013	2.0575	2 5 2		2.0961	152]		
210 { 042	2.0515	227	vs,vd	2.0891	125 }	s,vd	
321	2.0276	53	w	2.0713	33	vw	
330	1.9590	12	VVVW	2.0006	14 18	VVVW VVW	
312	1.8457	27	vw	1.8847	18	vvw	
$300 \begin{cases} 033 \\ 060 \end{cases}$	1.8124	121	m w	1.8461	219 95	s w	
f 213	1.7735	88]		1.8085	162]		
211 242	1.7697	90	s,vd	1.8040	149	vs,vd	
002 400	1.7492	256	s	1.7887	276	5	
152	1.7283	0	-	1.7603	5 18	-	
161	1.6839	8	vvvw	1.7148	19	vvw	
$102 \begin{cases} 411 \\ 420 \end{cases}$	1.6652	2		1.7022	1	-	
332	1.6626	17]		1.6967	16	vvw	
143	1.6126	0	-	1.6428	1	wd	
301 253	1.6041	9	vvvw	1.6345	25	" ,u	
350	1.5871	0	-	1.6191	1	-	
220 004	1.5653	13	vvw,d	1.5938	49	Ŵ	
112	1.5285	2	-	1.5617	20	w	
114	1.5184	10	-	1.5471	6	-	
170	1.5107	20]		1.5382	3		
310 \ 053	1.5094	20	w,vvd	1.5338	18	w,vvd	
071	1.5023	24		1.5294	15		
323 422	1.4980	13] 6]		1.5284	3	-	
202 1440	1.4693	9∫	VVW	1.5005	4	-	
221 204	1.4337	14	vvw,d	1.4615	2	-	
352	1.4168	11	vyvw	1.4449	5	-	
134 361	1.4115	10	-	1.4380	2	-	
510	1.3879	9]		1.4191	6	-	
$311 \begin{pmatrix} 224\\ 253 \end{pmatrix}$	1.3860	56 (58 (m,vd	1.412/	29	w,vđ	
271	1.3804	65,		1.4063	26		
172 1044	1.3616	181	• w	1.3865	24 31		
400 080	1.3538	25	vvw	1.3781	8	w,vđ	
343	1.3509	0	_	1.3777	3 J 12	-	
۲413 ۲413	1.3327	117	-	1.3606	71]	-	
212 442	1.3311	96 109	s	1.3587	53 } 64	m,d	
521	1.3245	21	vvvw	1.3538	7	-	
530	1.3048	1 7	-	1.3335	2 13	-	
314	1.2941	1	-	1.3199	0	-	
370	1.2893	- 4	-	1.3143	U	-	

*Indices of Fe₂P-like subcell

of the intensity measurements of the super-reflections, which are not superimposed in reciprocal space because of the body centered extinctions (Fig. 1). The reciprocal lattice points of the Fe₂P-like subcells of the three orientations of the trilling coincide. However, the orthorhombic distortion of the structure has little influence on the intensities of the subcell and thus the small errors arising through the coincidence of the three pseudohexagonal lattices of the subcell do not affect the structure refinement significantly. Additional data were obtained from an irregularly shaped specimen ($\sim 0.15 \times 0.10 \times 0.10$ mm), which contained nearly equal amounts of each orientation. This specimen was rotated around the pseudohexagonal [100] axis in a Weissenberg camera (multiple-film technique, Mo $K\alpha$ radiation). Three reciprocal lattice layers were evaluated. Only the non-coinciding superstructure reflections and the h00 reflections were included in the leastsquares refinement, although the intensities of the subcell (divided by 3, to account for the twinning) were utilized to obtain additional scale factors.

Structure determination and refinement

The conditions for possible reflections (*hkl* only with h+k+l=2n, hol only with h=2n and l=2n) lead to the two space groups $Ima2-C_{2v}^{22}$ and $Imam-D_{2h}^{28}$. It was previously mentioned that the intensity calculation for the Fe₂P-type subcell ($P\overline{6}2m$) gave very good agreement with the observed intensities. Therefore, it was concluded that the atomic positions in the four times larger orthorhombic cell would be essentially retained. This corresponds to 12 formula units of TiFeSi in the orthorhombic unit cell with a calculated density of $\rho = 5.51$ g.cm⁻³ as compared with the experimental density of q = 5.57 g.cm⁻³, given by Markiv *et al.* (1967). Since the sixfold inversion axis in the Fe_2P -type structure is incompatible with the twofold rotation axis required by Imam, this space group could be ruled out. During a least-squares refinement of the Fe₂P-like subcell it was found that the silicon atoms in TiFeSi correspond to the phosphorus atoms and the titanium atoms take the places of the iron atoms in position 3(g), of the Fe₂P-type structure (Rundqvist & Jellinek, 1959). There are two possibilities for the description of the Fe₂P-type subcell of TiFeSi in space group *Ima2*: one with Ti atoms at $x \simeq 0$ and $\frac{1}{2}$, the other with Ti at $x \simeq \frac{1}{4}$ and $\frac{3}{4}$. Only the latter could be refined by the least-squares method. A full-matrix program by Busing, Martin & Levy (1962) was used, giving unity weights to all observed structure factors. The values of the atomic scattering factors were taken from International Tables for X-ray Crystallography (1962) and were corrected for the real part of the anomalous dispersion (Dauben & Templeton, 1955).

During the first series of least-squares cycles the positional parameters of the atoms, their isotropic temperature parameters and one scale factor were refined. It should be emphasized that a refinement of the scale factors for individual reciprocal lattice layers was not feasible in this case, since the data for the superstructure and substructure were recorded on different films in the precession camera and the degree found for the orthorhombic distortion of the hexagonal subcell depends to a large extent on the ratio of scale factors for the sub- and superstructure. The incorporation of the Weissenberg data, recorded from the practically ideal trilling mentioned above, however, allowed double checks on the scale factors. All scale factors agreed within 10%.

The refinement of the isotropic thermal parameters gave values for the transition metal atoms varying between $B=0.32\pm0.10$ Å for Ti(3) and $B=0.62\pm$ 0.06 $Å^2$ for Fe(1). The corresponding values for the lighter silicon atoms were $B=0.74\pm0.14$ Å for Si(1) and B = 0.98 + 0.20 Å² for Si(2). These variations were thought to be of no physical significance and it was preferred to fix these parameters with the weighted averages of 0.5 for the transition metal atoms and 0.8 for the silicon atoms in the subsequent refinement cycles. As a test for possible disorder in the structure, occupancy factors were refined together with the positional parameters. The resulting values for occupancy factors varied between 0.98 ± 0.03 for Si(1) and $1.03 \pm$ 0.03 for Ti(3) with the exception of 0.93 ± 0.05 for Si(2) and 1.07 ± 0.03 for Fe(2). Therefore, substantial disorder can definitely be ruled out. There is a slight indication that either Si(2) has a somewhat larger thermal parameter or that this position has a small deviation from full occupancy.

In the final least-squares cycles only the positional parameters and the overall scale factor were varied, resulting in the positional parameters listed in Table 2. The final R value $(R = \sum |F_o - F_c| / \sum F_o)$ is 0.094 for the 344 observed structure factors (Table 3). Since this residual value depends mostly on the agreement of the strong subcell reflections, it cannot be taken as a measure for the accuracy of the orthorhombic distorted structure. However, the R value for the 210 observed superstructure reflections is similar: R = 0.096.

The structures of TiFeGe and TiCoGe

Alloy specimens of these two compounds were prepared in the same way as described for TiFeSi. The Guinier-de Wolff pattern taken with the double-film technique (to absorb fluorescent radiation) and Cu $K\alpha$ radiation showed no impurity lines for the two samples. The annealing treatment (7 days, 840°C in silica tubes) only resulted in a sharpening of the powder lines. The pattern of TiFeGe was very similar to that of TiFeSi, clearly showing the weak superstructure lines of the TiFeSi structure and could be indexed with a similar orthorhombic unit-cell. Least-squares refinement of the lattice constants (as described above) gave $a=7.155 \pm 0.002$, $b=11.025 \pm 0.007$, $c=6.405 \pm 0.003$ Å and $V=505.3 \pm 0.7$ Å³. The intensity calculation (Yvon, Jeitschko & Parthé, 1969) using the positional parameters of TiFeSi gave good agreement with the observed intensities (Table 1).

In the Guinier-de Wolff pattern of TiCoGe, on the other hand, no superstructure lines could be observed. It could be indexed with the hexagonal Fe₂P-type unitcell, giving least-squares refined lattice constants of $a = 6.222 \pm 0.002$, $c = 3.7267 \pm 0.0010$ Å, $c/a = 0.5990 \pm 0.0010$ 0.0002, $V = 124.94 \pm 0.12$ Å³. In the recent literature several ternary compounds with the Fe₂P-type structure have been reported. Some examples are given in Table 4, which also contains the distribution of atoms on the four equipoints of the Fe₂P-type structure. It can be seen that the large transition metal atoms always occupy positions 3(g), whereas the smaller transition metal atoms were found at either the 3(f) sites or at the positions 2(c) and 1(b). Therefore, the atomic distribution, as suggested by the subcell of TiFeSi, was not assumed to be necessarily correct. In order to decide which of the two ordering schemes applied for TiCoGe, relative line intensities were estimated with the aid of an intensity strip and were compared with the calculated intensities for these ordering schemes. The two positional x parameters were varied systematically in these calculations. Agreement between observed and calculated intensities could be obtained only for the atomic distribution and the positional parameters given in Table 4. Thus the atomic order in TiCoGe is the same as in the borides and silicides of Table 4 and corresponds also to the subcell of TiFeSi. A listing of observed and calculated powder intensities of TiCoGe is given in Table 5.

Discussion

A large number of phases have been ascribed the ordered Fe₂P-type structure in the recent literature. Kripjakevič, Markiv & Melnyk (1967) were the first

Table 2. Atomic parameters of TiFeSi and their standard deviations

			$\sigma(x)$		σ (y)		$\sigma(z)$	
	Position	x	(×104)	у	(×104)	z	(×104)	В
Ti(1)	4(<i>b</i>)	1 <u>4</u>	_	0.2004	7	0.2964	14	0∙5 Ų
Ti(2)	4(b)	4	_	0.7793	6	0.2707	14	0.2
Ti(3)	4(b)	4	_	0.9979	6	0.9178	15	0.2
Fe(1)	8(c)	0.0295	7	0.3764	4	0.1200*	_	0.2
Fe(2)	4(a)	0	-	0	_	0.2201	12	0.2
Si(1)	8(c)	0.0060	13	0.1675	9	0.9953	18	0.8
Si(2)	4(b)	14	_	0.9747	11	0.2055	23	0.8

* Held constant to fix the origin of the unit cell.

to report the ordering (Table 4) for the four equiatomic phases in the systems {Zr, Hf}-Ni-{Al, Ga}. Analogous positioning of the atoms has been reported for CeNiAl (Table 4) by Dwight, Mueller, Conner, Downey & Knott (1968), who found almost sixty more phases, all at the composition 1:1:1 in the ternary sys-

Table 3. Observed and calculated structure factorsof TiFeSi

h k l F _o F _c	hkl F _o F _c	h k l F F F	h k 1 F _o F _c	hkl F _F
0 2 0 55 38	1 11 2 39 33	2 4 4 87 93	3 11 4 27 21	7 15 0 <26 7
0 4 0 60 48	1 13 2 17 20	Z 6 4 133 115	3 13 4 21 19	7 17 0 <28 0
0 8 0 168 160	1 2 3 45 50	2 10 4 <18 10	3 2 5 60 58	7 21 0 30 34
0 10 0 115 113	1 4 3 12 8	2 12 4 171 155	3 4 5 45 39	7 2 1 76 78
0 12 0 <18 19	1 6 3 39 40	2 14 4 21 18	3 6 5 43 39	7 4 1 27 26
0 14 0 43 53	1 8 3 24 24	2 10 4 28 29	3 10 5 36 35	7 8 1 <18 10
0 3 1 85 65	1 12 3 44 44	2 3 5 99 104	3 12 5 22 22	7 10 1 57 61
0 5 1 286 250	1 14 3 46 43	2 5 5 64 69	3 14 5 65 55	7 12 1 32 35
	1 16 3 25 23	2 7 5 116 119	3 3 6 37 36	7 16 1 623 18
0 11 1 104 100	1 3 4 <8 6	2 11 5 36 36	3 5 6 18 16	7 1 2 59 62
0 13 1 94 102	1 5 4 30 30	2 13 5 <21 16	3 7 6 60 53	7 3 2 50 54
0 15 1 20 27	1 7 4 32 26	2 15 5 75 60	3 9 6 38 40	7 4 3 417 4
0 2 2 54 54	1 11 4 28 32	2 2 6 68 81	3 2 7 32 32	7 6 3 80 73
0 4 2 274 244	1 13 4 <13 8	2 4 6 87 99	3 4 7 21 23	7 3 4 55 54
0 6 2 83 72	1 15 4 13 16	2 6 6 < 18 9	3 6 7 49 53	7 5 4 56 57
0 10 2 95 92	1 4 5 39 40	2 10 6 29 31	3 1 8 56 53	7 6 5 45 49
0 12 2 85 89	1 6 5 24 26	2 12 6 44 29	3 3 8 <18 12	7 5 6 24 27
0 14 2 65 78		2 1 7 18 23	3 5 6 19 22	7 7 6 53 56
0 3 3 213 197	1 12 5 23 23	2 5 7 80 88	3 9 8 < 22 16	7 8 7 <23 15
0 5 3 102 110	1 14 5 40 34	2 7 7 59 60	3 2 9 24 25	7 7 8 < 23 15
0 7 3 94 93		2 9 7 51 48	3 4 9 26 24	8 0 0 180 179
0 11 3 <19 16	1 5 6 21 20	2 2 8 22 21	3 8 9 < 23 19	9 1 0 46 48
0 13 3 22 30	1 7 6 20 20	2 4 8 36 45	3 10 9 35 43	9 3 0 40 42
0 0 4 77 77	1 9 6 18 22	2 6 8 66 67	3 9 10 32 30	9 5 0 70 70
0 4 4 192 177	1 2 7 26 27	2 1 9 < 21 8	4 0 0 360 421	9 9 0 < 25 0
0 6 4 95 96	1 4 7 21 21	2 3 9 22 17	4 4 2 162 156	9 11 0 < 26 28
0 8 4 95 91	1 6 7 29 30	2 5 9 79 80	4 8 4 74 62	9 13 0 27 28
0 12 4 103 103	1 10 7 21 25	2 0 10 47 44	5 3 0 47 60	9 17 0 28 33
0 1 5 80 74	1 1 8 22 24	2 2 10 51 53	5 5 0 91 104	9190 < 300
0 3 5 104 119	1 3 8 26 23	2 4 10 77 72	5 7 0 33 38	9 2 1 46 50
0 7 5 76 75	1 7 8 21 24	3 1 0 56 73	5 11 0 22 30	9 6 1 53 59
0 9 5 93 92	1 9 8 22 23	3 3 0 54 58	5 13 0 27 35	9 A 1 < 24 22
0 11 5 32 38	1 2 9 20 19	3 5 0 < 14 3	5 15 3 51 60	9 10 1 53 57
0 2 6 79 82	1 6 9 28 30	3 9 0 86 87	5 19 0 28 9	9 14 1 27 39
0 4 6 104 96	1 8 9 < 21 19	3 11 0 30 34	5 21 0 <29 6	9 16 1 <27 22
0 8 6 100 101	1 9 10 <22 22	3 13 0 < 20 3	5 4 1 13 14	9 3 2 40 46
0 10 6 68.75	1 11 10 <23 12	3 17 0 <24 10	5 6 1 75 83	9 2 3 67 69
0 1 7 35 44	1 10 11 <23 9	3 19 0 37 55	5 8 1 28 32	9 4 3 <21 10
0 3 7 103 101	2 0 0 62 81	3 2 1 76 84	5 12 1 26 32	9 5 4 < 22 13
0 7 7 89 77	2 8 0 98 104	3 4 1 35 35	5 14 1 41 45	9 4 5 < 22 18
0 9 7 54 46	2 10 0 47 46	3 6 1 47 49	5 16 1 21 27	9 6 5 44 51
0 2 8 17 20	2 14 0 33 38	3 10 1 56 55	5 1 2 48 50	9 7 6 < 23 15
0 4 8 73 75	2 16 0 62 65	3 1 2 1 47 42	5 3 2 59 67	10 0 0 69 46
0 6 8 46 36	2 5 1 116 118	3 14 1 55 53	5 2 3 81 96	
1 5 0 85 84	2 9 1 126 119	3 18 1 <25 19	5 6 3 64 62	11 5 0 < 26 14
1 7 0 <9 5	2 11 1 109 107	3 20 1 < 25 3	5 3 4 35 41	11 7 0 27 30
	2 13 1 81 80	3 1 2 54 66	5 3 5 24 29	11 9 0 40 47
1 13 0 23 22	2 17 1 54 50	3 5 2 53 51	5 6 5 57 59	11 13 0 <28 10
1 15 0 40 42	2 4 2 129 123	3 7 2 72 62	5 5 6 40 43	11 15 0 <29 5
1 17 0 26 27	2 8 2 170 161	3 11 2 47 48	5 6 7 38 39	11 4 1 25 16
1 21 0 <29 19	2 10 2 73 68	3 13 2 36 34	5 8 7 25 30	11 6 1 30 36
1 2 1 21 19	2 12 2 74 76	3 15 2 38 31	5 7 8 35 39	
1 6 1 38 30	2 16 2 89 77	3 2 3 46 55	5 6 0 <24 17	11 12 1 25 24
1 8 1 32 34	2 18 2 24 18	3 4 3 <11 5	5 10 9 < 24 22	11 14 1 38 42
1 10 1 45 42	2 1 3 117 122	3 6 3 74 73	5 9 10 <24 22	11 1 2 32 34
1 14 1 34 35	2 5 3 125 119	3 10 3 66 62	6 4 2 64 66	11 2 3 32 38
1 16 1 29 28	2 7 1 165 158	3 1 2 3 37 36	6 8 4 54 44	11 4 3 < 23 2
1 18 1 <23 12	2 9 3 34 26	3 14 3 <24 9	7 1 0 67 70	11 3 4 26 33
1 1 2 8 8	2 13 3 24 31	3 1 4 17 16	7 5 0 21 24	12 0 C 81 72
1 3 2 46 46	2 15 3 39 31	3 3 4 54 50	7 7 0 45 48	14 0 0 76 51
1 5 2 7 8	2 17 3 69 53	3 5 4 64 64	7 9 0 71 76	16 0 0 35 29
1 9 2 36 32	2 2 4 118 119	3 9 4 26 27	7 13 0 <25 14	1
1	J		I	L

Table 5. Calculated and observed intensities of TiCoGe

Debye-Scherrer camera, Cu Ka radiation. In the low-angle range, data from a Guinier-de Wolff pattern were incorporated.

hk l	d _c (Å)	чc	1 ₀	hk1	d _د (Å)	I _c I _o
100	5.3884	3	2	410	1.1758	11 1.20
001	3.7267	8	8	321	1.1733	23
110	3.1110	19	20	312	1.1658	3 3
101	3.0651	6	6	113	1.1537	56
200	2.6942	3	2	203	1.1281	14 13
111	2.3882	56	50	411	1.1214	9 11
201	2.1834	100	100	402	1.0917	54
210	2.0366	32	30	500	1.0777	2 -
002	1.8633	29	25	213	1.0605	10 10
300	1.7961	23	20	330	1.0370	41,0
211	1.7872	39	40	501	1.0353	5∫10
102	1.7610	0	0	322	1.0301	2 -
301	1.6180	4	3	303	1.0217	1 -
112	1.5985	5	4	420	1.0183	1 -
220	1.5555	8	10	331	. 9990	1 -
202	1.5325	1	-	412	.9944	24 30
310	1.4945	3	4	421	.9823	13 15
221	1.4355	0	-	223	.9707	0 -
311	1.3871	7	5	510	.9678	2 2
212	1.3748	16	14	313	.9553	64
400	1.3471	4	3	511	.9367	22 20
302	1.2932	16	19	502	.9329	7), (
401	1.2669	0	-	004	.9317	11] 12
003	1.2422	0	-	104	.9181	a' -
320	1.2362	1	-	403	.9132	0 -
103	1.2105	1	-	332	.9061	26 25
222	1.1941	8	7			

tems {Y, R. E., Zr, Hf, Th, U}–{Fe, Ru, Co, Rh, Ir, Ni, Pd, Pt, Cu}–{Al, Ga, In}. Steeb & Petzow (1966) have reported that the phases U{Co, Ir, Ni} Al are of the Fe₂P-type. The assumed atomic order (U at the (P) positions, the other atoms statistically at the Fe positions), however, does not seem well substantiated. Lam, Darby, Downey & Norton (1967) have found UFeAl to be isotypic with UCoAl and UIrAl.

The atomic order in the phases NbFeB and TaFeB (Kuz'ma, 1967), NbMnSi (Deyris, Roy-Montreuil, Fruchart & Michel, 1968) and NbMnGe (Roy-Montreuil, Deyris, Fruchart & Michel, 1968) is different (Table 4). At first it may seem surprising that, for example, the Ni atoms in CeNiAl and ZrNiAl do not correspond to the Fe atoms in NbFeB. There are, however, structure-chemical arguments for the support of both ordering schemes:

(a) Both nickel and boron are the elements with the smallest atomic volume in their respective compounds; they are therefore suited for the phosphorus site in Fe_2P , which has the lowest coordination number.

Compound	3(g)	3(<i>f</i>)	2(c) and $1(b)$	Reference
Fe ₂ P	Fe, $x = 0.594$	Fe, $x = 0.256$	Р	1
ZrNiAl	Zr, $x = 0.600$	A1, $x = 0.256$	Ni	2
CeNiAl	Ce, $x = 0.580 \pm 0.001$	Al, $x = 0.219 \pm 0.00$	1 Ni	3
NbFeB	Nb	Fe	В	4
NbMnSi	Nb, $x = 0.589$	Mn, $x = 0.242$	Si	5
TiCoGe	Ti, $x = 0.57 \pm 0.01$	Co, $x = 0.245 \pm 0.01$	Ge	6

Table 4. Atomic order in Fe₂P-type structures (space group $P\overline{6}2m - D_{3h}^3$)

References

1 Rundqvist & Jellinek (1959).

2 Kripjakevič, Markiv & Melnyk (1967).

3 Dwight, Mueller, Conner, Downey & Knott (1968).

4 Kuz'ma (1967).

5 Deyris, Roy-Montreuil, Fruchart & Michel (1968).

6 This work.

(b) Although aluminum is chemically related to boron, silicon and germanium, it is known to behave sometimes like a transition metal. For example, it participates in the formation of such typically inter-transitionmetal structures as the σ -phase structure (σ -Nb₂Al and σ -Ta₂Al). This makes the correspondence of the Al atom in ZrNiAl with the Fe atom in NbFeB at least plausible.

(c) In alloys of the large transition metals of the third and fourth group of the periodic system, the elements of the eighth group do frequently occupy the boron positions in structure types, such as CrB, FeB and AlB₂ (for examples, see the compilation by Pearson, 1967) and thus have a coordination of nine nearest neighbours which is also the case for the Ni atoms in CeNiAl and ZrNiAl.

(d) The isotypy of many borides, phosphides, silicides and germanides, on the other hand, (Aronsson, Lundström & Rundqvist, 1965) supports the atomic order of NbFeB and TiCoGe.

The structure of TiFeSi can be derived from the ordered Fe₂P-structure by shifts in the atomic positions of up to 0.4 Å. The interatomic distances, however, are not as much affected by these shifts, since the positions of the neighbouring atoms are adjusted accordingly (Fig. 2). It must be emphasized that the weak superstructure-lines of TiFeSi and TiFeGe are barely visible on Debye-Scherrer films. It therefore seems



Nb5 Cu4 Si4

Fig.2. Projections of the structures TiFeSi, TiCoGe (ordered Fe₂P-type), TiNiSi (ordered PbCl₂-type, *E* phase; setting *Pnma*) and Nb₅Cu₄Si₄. In the projection of TiFeSi the atoms at $x=\frac{1}{4}$ and $x=\frac{3}{4}$ are connected by thin lines; atoms at $x\simeq 1$ (connected by thick lines) and $x\simeq\frac{1}{2}$ are superimposed. In the structures of TiCoGe, TiNiSi, and Nb₅Cu₄Si₄ atoms connected with thick and thin lines ar eseparated by $\frac{1}{2}$ translation period in the projection direction. Small open circles: Ti, Nb; solid circles: Fe, Co, Ni, Cu; large open circles: Si, Ge

possible that some of the ternary phases, reported to be isotypic with Fe₂P, actually crystallize with the TiFeSi structure. The Guinier-de Wolff pattern of a sample of ZrNiAl (prepared in the same way as TiFeSi), however, contained only lines of the small Fe₂P-type unit-cell, in agreement with the results reported by Kripjakevič *et al.* (1967). The interatomic distances in TiFeSi (Table 6) show a regularity which has also been observed for the structure of $Zr_4Co_4Ge_7$ (Jeitschko, 1969): all distances transition-metal-silicon (germanium) are shorter than the corresponding sum of the coordination number 12 radii, whereas all distances transition-metal-transitionmetal are longer (with the exception of two Ti-Fe dis-

Table 6. Interatomic distances D(Å) and coordination numbers (CN) in the structure of TiFeSi

The values are compared with the sum of the atomic radii for CN 12. All distances shorter than 3.9 Å are listed. The values given for CN are the numbers of neighbors within 1.15 $(r_A + r_B)$ and (in brackets) the number of additional neighbors within 1.30 $(r_A + r_B)$.

	Ti(1): CN	15(+2)			Fe(2): C	N 12	
	D	$r_A + r_B$	Δ		D	$r_A + r_B$	⊿
1Ti(3)	3.237 ± 13	2.92	+0.32	2Ti(3)	2.725 + 11	2.73	0.00
2Ti(1)	3.322 + 6		+0.40	2Ti(1)	2.803 ± 8		+0.07
1Ti(3)	3.355 ± 12		+0.44	2Ti(2)	2.965 + 7		+0.24
(2Ti(2))	3.509 ± 12		+0.59	2Fe(1)	2.691 + 8	2.54	+0.15
2 Fe(1)	2.685 ± 9	2.73	- 0.04	2Si(2)	2.390 ± 13	2.61	- 0.22
2Fe(1)	2.601 ± 9	2.0	-0.04	2Si(1)	2.420 ± 13	- • •	-0.19
2Fe(2)	2.803 ± 7		+0.07	(-)	- ·-· <u>-</u> ···		• ••
2Si(1)	2.003 ± 7 2.554 ± 13	2.80	-0.25	2Fe(2)	3.499 + 2	2.54	+0.96
2Si(1)	2.554 ± 15 2.574 ± 14	200	-0.23				,
1Si(2)	2.374 ± 14 2.775 ± 16		- 0.02		Si(1): CN	9(+1)	
151(2)	2775 10		0 02		D	$r_A + r_B$	Δ
	$T_{1}(2): CN$	15(+3)		1Ti(1)	2.554 ± 13	2.80	-0.25
	D	$r_A + r_B$	Δ	$1 \operatorname{Ti}(3)$	2.555 ± 12		-0.24
1Ti(3)	3.141 + 11	2.92	+0.22	1Ti(2)	2.557 ± 13		-0.24
2Ti(2)	3.207 ± 4		+0.29	1 Ti(1)	2.574 ± 14		-0.23
$1 \operatorname{Ti}(3)$	$3 \cdot 245 + 13$		+0.33	$1 \operatorname{Ti}(3)$	2.580 ± 12		-0.22
[2Ti(1)]	3.509 + 2		+ 0.29	$1 \operatorname{Ti}(2)$	2.582 ± 13		-0.22
2Fe(1)	2.751 + 8	2.73	+0.02	1Fe(1)	2.400 ± 12	2.61	-0.21
2Fe(2)	2.965 + 7		+0.24	$1 \operatorname{Fe}(1)$	2.413 ± 13		-0.20
2Fe(1)	3.123 + 9		+0.39	$1 \operatorname{Fe}(2)$	2.420 ± 13		-0.19
2Si(1)	2.557 + 13	2.80	-0.24	[1Si(1)]	3.415 ± 20	2.68	+0.74
1Si(2)	2.580 + 16		-0.22	//			
2Si(1)	2.582 ± 13		-0.22	1Si(1)	3.582 ± 20		+0.90
[1Si(2)]	3.217 + 16		+0.42	2Si(1)	3.616 ± 11		+0.94
[1~1(-)]				1Si(1)	3.629 ± 21		+0.95
	Ti(3): CN	15(+2)		1Si(2)	3.779 ± 16		+1.10
	D	$r_A + r_B$	Δ	1Si(2)	3.881 ± 16		+1.20
1Ti(2)	3.141 + 11	2.92	+0.22	• •	SID CN	0(+ 1)	
$1 \operatorname{Ti}(1)$	$3 \cdot 237 + 13$		+0.32		SI(2): CIN	9(+1)	
$1 \operatorname{Ti}(2)$	$3 \cdot 245 \pm 13$		+0.33		D	$r_A + r_B$	Δ
1Ti(1)	3.355 + 12		+0.44	1Ti(2)	2.580 ± 16	2.80	-0.22
[2Ti(3)]	3.499 ± 2		+0.58	1Ti(3)	2.604 ± 19		-0.50
2Fe(2)	2.725 ± 11	2.73	0.00	1 Ti(1)	2·775 ± 16		-0.05
2Fe(1)	2.782 ± 9		+0.02	[1Ti(2)]	3·217 ± 16		+0.42
2Fe(1)	3.010 ± 9		+0.58	2Fe(1)	2.340 ± 10	2.61	-0.27
2Si(1)	2.555 ± 12	2.80	-0.24	2Fe(1)	2.345 ± 12		-0.26
2Si(1)	2.580 ± 12		-0.52	2Fe(2)	2.390 ± 13		-0.22
1Si(2)	2.604 ± 19		-0.50				
				1Ti(3)	3·703 ± 19	2.80	+0.90
1Si(2)	3·703 ± 19		+0.90	2Si(2)	3·499 <u>+</u> 2	2.68	+0.82
	$E_{\alpha}(1) \cdot CN$	$12(\pm 1)$		2Si(1)	3·779 ± 16		+1.10
		12(+1)	4	2Si(1)	3·881 ± 18		+1.20
	D	$r_A + r_B$	4				
1Ti(1)	2.685 ± 9	2.73	-0.04				
1Ti(1)	2.691 ± 9		-0.04				
1Ti(2)	2.751 ± 8		+0.02				
1Ti(3)	2.782 ± 9		+0.05				
1Ti(3)	3.010 ± 9		+0.28				
1Ti(2)	3.123 ± 9		+0.39				
1Fe(2)	2.691 ± 8	2.54	+0.15				
1Fe(1)	2.709 ± 10		+0.17				
[1Fe(1)]	3.086 ± 11	a	+0.55				
151(2)	2.340 ± 10	2.61	-0.27				
1Si(2)	2.345 ± 12		0.26				
1Si(1)	2.400 ± 12		-0.21				
1Si(1)	2·413 ± 13		-0.50				

tances which are only slightly shorter). A comparison of the interatomic distances in TiNiSi (Shoemaker & Shoemaker, 1965) and Nb₅Cu₄Si₄ (Ganglberger, 1968) shows the same regularity.

The coordination polyhedra for the corresponding atoms in these structures (Fig. 2) are very similar. Each large transition-metal atom (excepting the fifth Nb in Nb₅Cu₄Si₄) has 15 nearest neighbors: 10 transitionmetal atoms and 5 silicon or germanium atoms. These tightly bound silicon or germanium atoms form a more or less regular quadratic pyramid in each of these structures. The small transition-metal atom has 12 neighbors: 8 atoms at the corner of a quadrangular prism and 4 more atoms outside the prism side faces. The four tightly bound Si(Ge) atoms of this polyhedron form a tetrahedron in the structures of TiFeSi, TiCoGe and TiNiSi. In Nb₅Cu₄Si₄ one of these four Si neighbours is replaced by a Cu atom at a larger distance. Finally, the environment of the silicon (germanium) atoms is the same in all these compounds: nine transition-metal atoms at a rather short distance, with 6 atoms at the corners of a trigonal prism and 3 atoms outside the rectangular faces of the prism.

The structures of TiFeSi and TiCoGe on the one hand and TiNiSi on the other hand are even more related. The main difference in nearest neighbour environment in these structures arises through the nonequivalence of the Si(Ge) atoms in TiFeSi(TiCoGe): the one Si(Ge) atom in equipoint 8(c) [2(c)] has 6 Ti and 3 Fe(Co) atoms and the other in equipoint 4(b) [1(b)] has 3 Ti and 6 Fe(Co) atoms as nearest neighbors. On the average each Si(Ge) atom has five large and four small transition-metal neighbors as is also the case for the silicon atom in TiFeSi.

The close resemblance of the nearest neighbor environments in these phases is an indication of the stability of these polyhedra. The contraction in the transition-metal-silicon (germanium) distances suggests a strong bond between these elements. The bonding between the transition-metal atoms seems to be of less importance, as is indicated by the stretching of the corresponding distances.

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